

## Concentrations et ppm

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$$1 \text{ cm}^3 = 1 \text{ mL} = 1 \text{ g}$$

$$1 \text{ ppm} = 1 \text{ partie par million} = \frac{1 \text{ g}}{1\,000\,000 \text{ g}} = \frac{1 \text{ g}}{1\,000\,000 \text{ mL}}$$

$$1 \text{ ppm} = \frac{1 \text{ cm}^3}{1 \text{ m}^3}$$

$$1 \text{ ppm} = \frac{1 \text{ mg}}{1 \text{ kg}}$$

$$1 \text{ ppm} = \frac{1 \text{ mg}}{1 \text{ L}}$$

$$\% \text{ m/V} = \frac{\text{masse (g)}}{\text{Volume (mL)}} \times 100$$

$$\% \text{ m/m} = \frac{\text{masse (g)}}{\text{masse (g)}} \times 100$$

$$\% \text{ V/V} = \frac{\text{Volume (mL)}}{\text{Volume (mL)}} \times 100$$